

On the Theory of Conversion of Solid Ammonium Halides at
Low Temperatures; I. Ammonium Chloride, by Takeo
Nagamiya.

~~per:~~
JAPANESE, Proc. Phys. Math Soc Japan, Vol XXIV, 1942,
~~pp 137-164.~~ pp 137 - 164.

ANC Tr 649

Sci - Chemistry

AS 91

Oct 1951 GTS

On the Theory of Conversion of Solid Ammonium Halides
at Low Temperatures; II. Ammonium Bromide, by
Takeo Nagamiya.

JAPANESE, per, Proc. Phys. Math Soc Japan, Vol XXV,
1943, pp 510 - 532.

AEC Tr 650

SLA
2484

Sci - Chemistry

A8292

Oct 1951 CTS

Studies on Feeding Standard of the National Safety
Force. Report No IV. Basal Metabolism, Relative
Metabolic Rate for Various Workloads, and Energy
Consumption for Infantry Personnel at Camp Korima,
by Etsu Hattori, Miura Shogo Shinjiro Suzuki, Shoji
Kawada, Shinichi Nagasawa, Tatsuro Kaga, Kiyoie
Yamamoto, Shunichi Ashima, 23 pp.

JAPANESE, per, Hosen Eihei, Vol I, No 5, Oct 1958,
pp 7-17.

REF 7-33

Ref - Medicine
Aug 59

93, 197

Study on Automatic Computers in Nippon Electric Co,
by K. Haganari. UNCLASSIFIED

JAPANESE, per, Denki Tsushin Gakkaishi.

USASIA Tr 972R

Sci - Electron

Feb 58

59,240

Prevention and Treatment of Clonorchiasis,
by K. Nagano, 8 pp.

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JAPANESE, ~~yes~~, Even Saishin Kaseichu Byodoku,
(Recent Parasitology), Edited by Shoru Morishita,
(Tokyo: Igakusha-in, 1951), ~~See~~, pp 1-43,
(Selected Parts Only). Vol IV

HYR 9-22

Scientific - Medicine
Nov 56 GTS

40,391

Prevention and Treatment of Clonorchiasis, by
Kenji Hagan, 8 pp.

JAPANESE, par, Kenji Hagan, Vol. IV, 1951, pp
1-43.

SLA Tr 1501/56

45, 250

Sci - Med
Mar 57 CIB

62-23092

Nagano, Masamitsu, Takeshima, Kenji, and Otsuki,
Ryuichi.

INTRINSIC VISCOSITY AND SPINNING OF HEAT-
TREATED PVA. Pt. 3 of Study of PVA Fibers.

[1962] Ill. p. 5, 6th.

Order from SLA \$11.60

62-20092

I. Nagano, M.
II. Takeshima, K.
III. Otsuki, S.
IV. Title Study...

Trans. of [Kobunshi Kagaku] (Japan) 1955, v. 12,
p. 79-85.

DESCRIPTORS: *Polyvinyl alcohol, *Fibers (Synthetic),
Heat treatment, Viscosity, Solutions.

The heat-treated PVA solution is in a different state of
solution from that obtained from PVA not so treated.
It was found that the $[\eta]$ of Huggins equation increases
with the progress of heat-treatment, whereas
this value decreases when treated excessively. On
the other hand, the b' value decreases with the prog-
(Materials-Textiles, TT, v. 9, no. 8)

Office of Technical Services

Aging of Viscose, by Masamitsu Nagano, 3 p.

JAPANESE, patent, Showa 28-5368, 1953, request
of Official Patent rept. 19 Oct 53,
Presentation of Request 21 Nov 51, Spec. appl.
No Showa 26-15490, by Kuratoki Rayon Company,

SLA 57-17843

Sci
Mar 60
Vol 2, No 11

109,619

Researches on Gas Turbines Experimental N^o 20 Turbo-
Jet, by Dr. Osamu Nagano, 24 pp. (AP 648561)

JAPANESE, bk, 1953, pp 12-40, 148-158.

ATTN: F-73-8517/V

Sci - Engineering
Feb 1957 CTS/DEI

44, 383

GAS FLOW IN A SLANTED TOP-BOTTOM I.D FURNACE, BY
H. HASHIMOTO, Y. NAGANO.

JAPANESE, PER, TETSU TO MAGNET, VOL XLIV, NO 9,
1958, PP 1055, 1056.

HB 5621

SCI - ENGR
OCT 62

212,092

FLOW IN A SPHERICAL BUBBLE OXYGEN LD FURNACE,
BY H. HASHIMOTO, Y. NAKANO.

JAPANESE, PER, TETSU TO HAGANE, VOL XLV, NO 3,
1953, PP 248, 249.

NO 3620

SCI - ENGR
OCT 62

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Central of Japan in the Middle East in
 Japanese Encyclopedia, by Y. Nagano, W. Kikuchi.
 J. Kikuchi, ed., Tokyo, 1970, Vol. 45,
 pp. 210-211.
 (J. Kikuchi)

Y. Nagano

314,5 11

Sec - Mechanical, Industrial,
 Civil, and Marine Engineering
 (J. Kikuchi)

Treatment of Some Metals With Low Ultrasonic Energy, by Y. Nagano, K. K. Kikuchi

JAPANESE, per, Tetsu To Hagane,
Vol. XLVI, No. 10, 1960, pp. 1000-1001

AB 5410

Sci. - Electron
Dec 61

On the Distribution of Oxygen in Molten Steel
During Refining, by K. Sugami, M. Igarashi, Y.
Nagano. UNCL

JAPANESE, per, Steel Met, Vol X, Jan 1948,
pp 1-4.

BRITISH IRON AND
STEEL IND 1073

Sol - 124/1073
Jan 59

79,586

* A Measurement of Neutron Slowing-Down Area
in Light Water, by T. Hatanaka, S. Ogura, M.
Shimizu, H. Aisu, T. Kondo, 9 pp.

JAPANESE.

AEC Tr-4322

Sci - Phys
Apr 61

149,067

A Measurement of Neutron Slowing-Down Area in
a Light Water, by T. Nagamura, S. Oguro, M.
Shimizu, H. Aisu, T. Kondo,

JAPANESE.

*AIX

Sci - 1961
8 Mar 61
List 50

On the Measurement of Exhaust Gas Density, by
Hiroo F. Tagajo, K. Tanabe.

JAF/WHSE, Lecture Delivered at the 9th Meeting
of the Japanese Institute of Mechanical
Engineering Kansai Branch 16th March 1939.

NLL M. 3626

Sci. - Engng

Aug. 68

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Fuel Injection and Combustion in the
Swirl Chamber of a Diesel Engine, by P. Nagao,
H. Kikimoto.

GERMAN, 1959, Motortechnik Z., 1959,
Hpp 303-305.

CHINO

207,603

Bo1 - 1887
Jul 62

Thermal Decomposition of Polyacrylonitrile
(Production of Hydrocyanic Acid), by Hideo
Hagao, Maruya Uchida, Koyu Yamaguchi,
8 pp.

JAPANESE, par. Kogyo Kagaku Zasshi,
Vol. XLIX, No. 6, 1956, pp 691-700.

SLA 60-18460

Sci
Vol. IV, No. 11
Jan 62

199, 248

Thermal Decomposition of Acrylonitrile-
Vinylidene Chloride Co-Polymers, by
Hideo Nagae, Nobuhisa Uchida, Teruo Yamaguchi,
1971

J.P.A.M.S., 1971, Tokyo Kagaku Zasshi, Vol 59,
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S.A. 111-104-30413

Hideo Nagae

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111,944

Effects of Added Reagents on the Aqueous Polymeri-
zation of Acrylonitrile. Study With An Electron
Microscope, by H. Wagon, M. Ishida.

JAPANESE, par, J Chem Soc Japan, Ind Chem Sect,
Vol. LXI, No 4, 1958, pp 465-469.

Assoc Tech Serv JJ-1464
" " " 99K24J

Sci - Chem

Apr 59

83,599

HAGATA, Tatsuo
HAGATA, Ippai

The Effect of various fractions of Typhoid Bacillus
filtrate upon the Yoshida sarcoma. 6-Pages.

Japanese, Gann, II, Nos. 2,3,4.
1949, 142-144

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Translated at the
National Institutes of Health
Bethesda, Md.

1-12-54

Observation of Generation and Decay of Supersonic
Wave Cavitation, by Junichi Sanoyoshi, Motoyoshi
Okushima, Nago Tajiri, 2 pp. UNCLASSIFIED

JAPANESE, rpt, Main Nihon Onkyo Gakkai Koen Ron-
bunshu, May 1960.

NAVY 2500/NRL 766

Sci - Physics
May 60

CIA/TED K-4091

116,217

Contribution to Improved Combustion in Ante-
Chamber Diesel Engines, by F. Egges,
Kaiserslautern.

GERMAN, par, Motortechnische Z., Vol XVIII,
1957, pp 301-306.

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Sed. + Engg
May 62

196, 615

Report of the Fur Seal Investigation in 1953, by
F. Nagasaki.

JAPANESE, rpt, Report of the Fur Seal Investigation
in 1953, No 2, 1953, pp 157-159.

Fisheries Res Board of Canada
Biol Station
Nanaimo BC B.C.

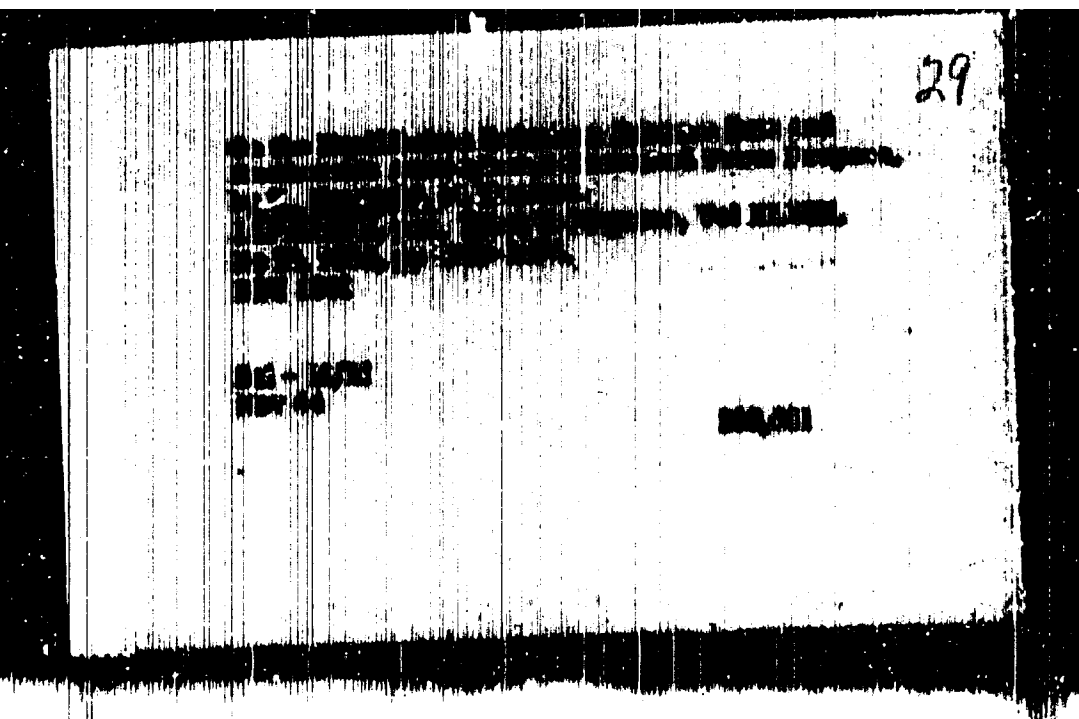
Sci

Feb 64

51
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Siliconizing of brass with Silicon Tetra-
chloride and H_2 , by K. Nagasaki.
JAPANESE, por, J. Japan Inst. Metals, 24
1960, pp. 28-32.
*JAS IT 70-57991

Sci/Engr
Aug 73

K. Nagasaki



61-10004

Naganaka, Noboru and Miyazaki, Shozo.
ON DECOMPOSITION BY TUNGSTEN. Rept. 1 of
Studies on the Catalytic Decomposition of Ammonia
Gas. [1960] 7p. 4 refs.
Order from S.L.A. m\$1.80, ph\$1.80 61-10004

Trans. of [Nihon] Kagaku Zasshi (Japan) 1949, v. 70,
no. 4, p. 134-136.

Decomposition reactions at 700° to 770°C for an initial
gas pressure range of 100 to 200 mm Hg showed that
the velocity is expressed most aptly by the equation of
C. N. Kurnian and E. S. Lerner (Phil. Mag. 7 10:
1015, 1930):

$$\frac{1}{v} = \frac{1}{k(T)} + \frac{g(T)}{k(T)} \frac{P_{H_2} \cdot M_{NH_3}^{1/2}}{P_{NH_3} \cdot M_{H_2}}$$

(See also 61-10002)

(CABOTAGE - PAYMENT, TT, v. 5, no. 2)

1. Ammonia--Decomposition
2. Tungsten--Catalytic properties

- I. Naganaka, N.
- II. Miyazaki, S.
- III. Title Studies...

Source of Technical Information

Studies on Uranium-Molybdenum Alloy, by Masayuki
Kawasaki, Ryokichi Hasegaki, Motono Itagaki,
Tatsuya Takemura,

JAPANESE, per, J Atomic Energy Soc Japan, Vol II,
1960, pp 136-140.

ABC 71-4466

Bo1 - Min/Met
May 61

150,868

Study on Corrosion Prevention Method Around
Splashing Zone for Marine Structures (Steel),
by S. Nagasaki,
JAPANESE, per. Doboku Gakkaishi, Vol. 21, No. 9,
pp 61-75.
Dist 0144

Sci/Mnt
Apr 70

S. Nagasaki

RECENT Construction Work of Tac-Man Beacon Light,
by Sakaji Nagasaki.

JAPANESE, per, Doboku-Onkai-Shi, Vol XLVI, No 3,
1961, pp 5-10.

REEL, F: Belvoir T-1579

Sol - Kogr
May 62

145, 951

Joint Toxic Action of Mixtures of Malathion
with Dimethoate and Dibrom on Adults of the
Common House Fly, *Musca Domestica* Vicina. by
S. Nagasawa, Y. Tsuruoka.

JAPANESE, per, Botyu Kagaku, Vol XXVII, No 3, 1962,
pp 78-81.

GB/52

Sci-Biol
July 63

236,310

The Influence of Salmon Eggs and Fry by Predaceous
Fishers, by Toyosumi Mikita, Ariaki Nagasawa.

JAPANESE, per, Scientific Reports Hokkaido Salmon
Hatchery, No 15, 1961, pp 69-83.

Biological Lab
Bur of Commercial Fisheries
Seattle, Washington

Sci Biol

198, 324

Mar 62

Organic Phosphorus Insecticide Composition,
by Masao Nagasawa, Takao Tsuboi, Kan.
Shinohara, Opp.
JAPANESE, Patent 6199/1963, cl. 30-F-37:
appl. no. 170/1961, 6 Jan 61, pub. 16 May
63, by Ihara Noyaku Co., Ltd.
CFFPI HA Code-P-312

Masao Nagasawa
33,19

Sel - Chemistry
Apr 67

Nagasawa, Masao, Yamamoto, Fukuro, and
Imamiya, Yoji.
AGRICULTURAL FUNGICIDES OF ARSENIC-CONTAINING ETHYLENE-BIS-DITHIO-CARBAMATE
SERIES. Feb 64, 10p
Order from SA \$22.00

SA Code-P-235

Trans. of published Japanese patent 7750/1959,
cl. 30-F-87 (30-F-91) appl. 9086/1956, 5 Apr 56,
pub. 3 Sep 59, by Iorihara Agricultural Chemicals
Co., Ltd. (Abstract available)

DESCRIPTORS: *Fungicides, *Arsenic compounds,
*Carbamates, Sulfur compounds, Ethylenes,
Preparation.

The invention relates to agricultural fungicides having
the purposes described in the specification, charac-
terized by having one kind or two or more kinds of
organic arsenic compounds as effective components
(Chemistry--Organic, TT, v. 11, no. 9) (over)

TT-64-1257

1. Title: Ethylene-bis-
dithiocarbamic acid
arsenic derivatives

I. Nagasawa, M.

II. Yamamoto, F.

III. Imamiya, Y.

IV. Patent (Japan)

pub. 34-7750

V. SA-Code-P-235

VI. Seisaburo Aoki,
Fujisawa (Japan)

Office of Technical Services

Nagasawa, Munetoshi and Taguchi, Taeko.
SHIKIMIC ACID IN FLUE-CURED TOBACCO LEAF.
[1961] (p. 12 refs.
Order from S.L.A. \$1.20

61-20211

Trans. of [Rihon Seibai Koshu Chuo Kenkyusho
Kenkyu Hokoku] (Japan) 1958, no. 99, p. 8-9.

DESCRIPTORS: *Tobacco, *Plants, *Shikimic acid.

The occurrence of shikimic acid in flue-cured tobacco leaf was confirmed by the results of the qualitative tests. The spectrophotometric analysis indicated its concentration to be approximately 16 mg. per 100 g. of dry material. The experimental results support the view that shikimic acid is an intermediate compound in the formation of chlorogenic acid during the curing of tobacco leaf. (Author)

(Agriculture--Plant Cultivation, TT, v. 7, no. 9)

61-20211

I. Nagasawa, M.
II. Taguchi, T.

206055

Office of Technical Services

Study of Tungsten Grains which Possess High
Electric Constants (Second Report), by Maruyoshi
Nagamura. 86

JOURNAL, for, Jushi Kagaku, Vol 23, 1948,

pp 57-59.

ADSI J-5176

ID AEC-035007

Maruyoshi Nagamura

Sci-Electronics

Nov 67

343,988

Study of Tungsten Oxides which Possess High
Piezoelectric Constants (First Report) by Morioka
Kiyomasa.

JAPANESE, per, Radio, Osaka, Vol 16, 1968,

pp 13-15.

ACR 3-1876

ID 2204075967

Radio, Osaka

Sci. Electronics
Nov 67

343,947

Nagawawa, S. and Funakubo, M.
MICROANALYSIS OF LEAD, ESPECIALLY TETRA-
ETHYL LEAD, IN AIR. [1963] Sp (figs omitted) 1 ref/
Order from S.L.A. \$1.10 TT-63-18950

Trans. of [Nihon Kagaku Zasshi] (Japan) 1948, v. 69,
p. 16-17.

DESCRIPTORS: *Lead, Microanalysis, *Ethyl radicals,
*Lead compounds, Air, Chemical analysis.

(Chemistry--Analytical, TT, v. 11, no. 5)

TT-63-18950

1. Title: Tetraethyl lead
- I. Nagawawa, S.
- II. Funakubo, M.

Office of Technical Services

(FDD 20885)

Dielectric Properties of Oxides of Rare Metals, by
Shigeyuki Nagasawa, 8 pp.

JAPANESE, per, Denki Kagaku, Vol XVIII, Jan 1950.

CIA/FDD/O-6115

FE - Japan

Sci - Physics

Jul 54 ODS

17,236

The Theory of Film Electro-Potential Differences of
High Molecular Films, (Part 2) The Relation Between
the Transfer Rate and Concentration, by M. Nagasawa,
7 pp.

JAPANESE, per, Chem Soc of Japan, Vol LXI, No 3,
pp 45-47.

S.I.A. Tr 504/56

Sci - Chemistry

36,984

Aug 1956

Comparison of Snow by Static and Kinetic Loads,
by Makoto Hagiwara, 8 pp.

JAPANESE, per, Seppyo, Vol V, 1943, pp 249-256.
CIA D 301883

120 120 120
OCE Tr 35

Scientific - Geography
Engineering

31/69
Feb 56 CTB/DEX

Studies on the "Breeding" of Canned Crab Meat, 61
by Y. Nagatsuma.

JAPANESE, par, Memoirs of the Faculty of
Hokkaido University, Vol VIII, No 2, Dec
1965.

Dept of Interior
Fish and Wildlife Service
BCE, Bureau of Foreign Fisheries

Y. Nagatsuma

Sci-

Aug 67

338,672

Nagase, H., and Sakaguchi, K.
ADDITION REACTION OF ETHYLENE OXIDE TO
LAURIC ACID. [1962] DR p. 14 refs.
Order from SLA 50 60 62-14288

Trans. of Kyoto Kagaku Zasshi (Japan) 1961, v. 64
[no. 6] p. 1005-1010

DESCRIPTORS: Ethylene oxide, Additives, Chemical
reactions, Fatty acids, Catalysts, Laurates

Sodium metal was used as the catalyst in the reaction
of ethylene oxide at atmospheric pressure by the
bubbling method, and under pressure by measuring the
rate of addition. The product analyses were done
after extracting the product with ether and determining
the acid value, saponification value, and hydroxyl value
of the extract. In both cases the addition curve shows
an inflection point when the average number of moles,
v, of ethylene oxide is one. For the products show dif-
(Chemistry: Physical, ET, v. 6, no. 4) (over)

62-14288

T. Nagase, K.
H. Sakaguchi, K.

62-14288

Office of Technical Services

Nagase, K. and Sakaguchi, K.
ETHYLENE OXIDE ADDITION REACTIONS TO
LAURYL AMINE. [1962] [16] p. 11 refs.
Order from SLA \$1.60

62-14289

Trans. of Kogyo Kagaku Zasshi (Japan) 1961, v. 64
[no. 6] p. 1031-1034.

DESCRIPTORS: *Ethylene oxide, Catalysts, Chemical
reactions, *Amines, Additives, Polymerization.

The addition of ethylene oxide to lauryl amine can be
divided into two reaction types depending on the kind
of catalyst used. One is the acid catalyzed reaction
(A.C. type), and the other is the base catalyzed reac-
tion (B.C. type). When the average number of moles,
 $\bar{v}$, of ethylene oxide is less than two, both reactions
can be clearly classified into these two types. These
reactions were studied in order to investigate the re-
action mechanism by examining the relationship of the
(Chemistry-Physical, TT, v. 8, no. 3) (over)

62-14289

I. Nagase, K.
II. Sakaguchi, K.

112013

Office of Technical Services

K. NAGASE

Studies of Addition Reaction of Ethylene Oxide.
Part VII. By-Products in Addition Reaction of
Ethylene Oxide, by K. Nagase
JAPANESE, per, Kogyo Kagaku Zasshi, Vol 64, 1961,
pp 1199-1203
NTC 71-15514-070

feb 72

62-14287

Nagase, K. and Sakaguchi, K.
ADDITION REACTIONS OF ETHYLENE OXIDE
UNDER PRESSURE. [1962] [15] p. 7 refs.

Order from SLA \$1.60

62-14287

I. Nagase, K.
II. Sakaguchi, K.

Trans. of Kogyo Kagaku Zasshi (Japan) 1961, v. 64
[no. 6] p. 1043-1047.

DESCRIPTORS: *Ethylene oxide, Additives, Chemical
reactions, Pressure, Temperature, Catalysts,
Polymerization, Solubility.

012011

Ethylene oxide was added to lauryl alcohol and nonyl
phenol under pressure (gauge pressure 1.5, 5.0 and
10.0 kg/cm²) at temperatures 100°, 125°, 150° and
175°C and catalyst concentration 0.25 to 2.0 mole
percent. The addition reaction curves, i.e. rates and
the distribution of the degree of polymerization were
investigated. The addition curve is a straight line
corresponding to the reaction time in the case of
(Chemistry--Physical, TT, v. 8, no. 5) (over)

Office of Technical Services

Investigation from 70 (10) and 10. Argentinean in
Hoping, by General MONTA, Southern MONTA, 23 pp.
JANUARY, 194, National Technical Report, Vol 10,
No 2, 194, pp 10-10. Economic Research
Industrial Co. 1944
AND VOL. 10-194

Kaneoni NACLASE

15.01

Sgt. - Nuclear Defense
Aug. 67

Separation of Polyethylene Glycol from the
Nonionic (Surface) Active Agent by the Method
of Counter-Current Distribution Extraction,
by Kinishiko Nagase, Kakei Sakaguchi, 11 pp.

JAPANESE, per, Kagyo Kagaku Zasshi, Vol LXIV,
I No 4, 1961, pp 635-638.

I SLA 61-20681

Sci
Mar 62
Vol VII, No 3

188, 095

61-10682

Nagase, Kunihiko and Sakaguchi, Kagen.
(DISTRIBUTION) OF PRODUCTS OF ADDITION OF
ETHYLENE OXIDE TO LAURYL ALCOHOL AND
NONYLPHENOL. [1960] [22]p. 67 figs. (omitted)
16 refs.

Order from SLA \$2.60

61-10682

Trans. of Kyoto Kagaku Zasshi (Japan) 1960, v. 63,
no. 4, p. 588-592.

DESCRIPTORS: Ethylene oxide, *Alcohols, *Phenols,
Nonyl radicals, Chemical reactions, Statistical
distributions.

1. Nagase, K
2. Sakaguchi, K.

Office of Technical Services

(Chemistry--Physical, TI, v. 6, no. 1)

Nagase, Shunji.

FLUORINATION OF MALEIC ANHYDRIDE BY

FLUORINE GAS. [1962] 6p. (2 figs. 2 tables

9 Japanese refs. 1 formula omitted) 2 refs.

Order from SLA \$1.10

62-16185

62-16185

I. Nagase, S.

Trans. of Kogyo Kagaku Zasshi (Japan) 1961, v. 64,
no. 8, p. 1400-1403.

DESCRIPTORS: *Maleic anhydride, Fluorination,
*Fluorine, Gases, Jets, Towers (Chemistry).

(62-16185)

In the direct fluorination of maleic anhydride with
fluorine by means of an improved jet reactor, per-
fluoropropionyl fluoride was obtained at a fairly high
yield rate through successive operations of the re-
actor. This suggests a new method of perfluoro-
propionyl fluoride production. It became apparent
that the use of said reactor makes maleic anhydride
(Chemistry--Organic, TT, v. 8, no. 5) (over)

Office of Technical Services

Nagase, Shin-ichi, Baba, H., and Kojima, R.
 PREPARATION OF PERFLUORO CARBOXYLIC
 ACIDS BY ELECTROCHEMICAL FLUORINATION OF
 UNSATURATED COMPOUNDS. [1962] 8p. (8 tables)
 6 refs. printed 2 refs.
 Order from S. A. 9, 10

62-16150

I. Nagase, S.
 H. Baba, H.
 R. Kojima, R.

Trans. of Electrochemical Society, 1962, 109, 1, 1-6.
 ap. 1, p. 1-6.

RESEARCH TOPIC: Carboxylic acids, Aromatic. Fluorination.
 Synthesis. Electrochemical fluorination. Aromatic
 Fluorination.

Experiments have been made to obtain a high yield of
 perfluorinated carboxylic acids from unsaturated compounds by the
 electrochemical fluorination of these compounds. Good yields of
 perfluorinated carboxylic acids have been obtained from
 unsaturated compounds. The results are discussed in terms of
 the mechanism of the electrochemical fluorination reaction.
 (Chemistry (Physical), IT, v. 8, no. 9) (600)

C 25567

Source: The National Archives

Nagase, Shunji, Baba, Hajime, and Kojima, Rimpet.
PREPARATION OF PERFLUOROCARBOXYLIC ACID
FROM KETONES BY ELECTROCHEMICAL
FLUORINATION. [1962] 8p. (5 tables omitted) 10 refs.
Order from SLA \$1.10 62-14642
Original copy poor quality; available on loan also

Trans. of [Kogyo Kagaku Zasshi] (Japan) 1961, v. 64,
no. 12, p. 2126-2128.

DESCRIPTORS: *Electrochemistry, Fluorination,
*Carboxylic acids, *Fluorides, Synthesis, *Ketones.

The yield of perfluorocarboxylic acid obtained from
ketones by the electrochemical fluorination was
excellent. Trifluoroethanoic acids are obtained from
acetone, acetyl acetone and ethyl acetoacetate. The
mixture of trifluoroethanoic and perfluoropropionic
acid were obtained from methyl ethyl ketone and
diethyl ketone. The mixture of perfluoropropionic
(Chemistry-Physics, TT, v. 8, no. 9) (over)

62-14642

I. Nagase, S.
II. Baba, H.
III. Kojima, R.

0255928

Office of Technical Services

62-14874

Nagase, T. and Kozima, I.
PREPARATION OF PERFLUOROCARBOXYLIC ACIDS
BY ELECTROCHEMICAL FLUORINATION ALCOHOLS.
[1962] 10p. (2 figs. 2 tables 11 refs. omitted).
Order from S.I.A. \$1.10 62-14874

T. Nagase, T.
H. Kozima, I.

Trans. of Kogyo Kagaku Zasshi (Japan) 1961, v. 64,
[no. 8] p. 1397-1400.

DESCRIPTORS: *Carboxylic acids, Acetic acids,
Propionic acids, Fluorides, Preparation, *Electro-
chemistry, *Alcohols, Ethanols, Propanols, Butanols,
*Fluorination.

C227656

Relatively high yields of perfluorocarboxylic acids have
been obtained by electrochemical fluorination of alcohols.
About 30 g. of ethyl alcohol, n-propyl alcohol, isopropyl
alcohol, or n-butyl alcohol was dissolved in 1 l. of
anhydrous hydrogen fluoride, and electrolyzed at various
(Chemistry--Organic, I.T. v. 9, no. 4) (over)

Office of Technical Services

Nagashima, Masao, Hide, Itta, Muroyama, Tatsuura,
and Watanabe, Susumu.

**NEW RESIN COMPOSITE WITH GOOD HARDENING
PROPERTY. [1963] 10p**

Order from OTS or SLA \$1.00

63-18906

Trans. of published Japanese patent 37 Showa (1962) -
2987 [cl. 35 H 72 (25 C 111) announced 30 May 62,
appl. no. 34 Showa (1959) - 24730, appl. 4 Aug 59, by
Sumitomo Bakelite Co., Ltd.

DESCRIPTORS: *Ambo plastics, *Thermosetting
plastics, *Epoxy plastics, Hardening, Addition
reactions.

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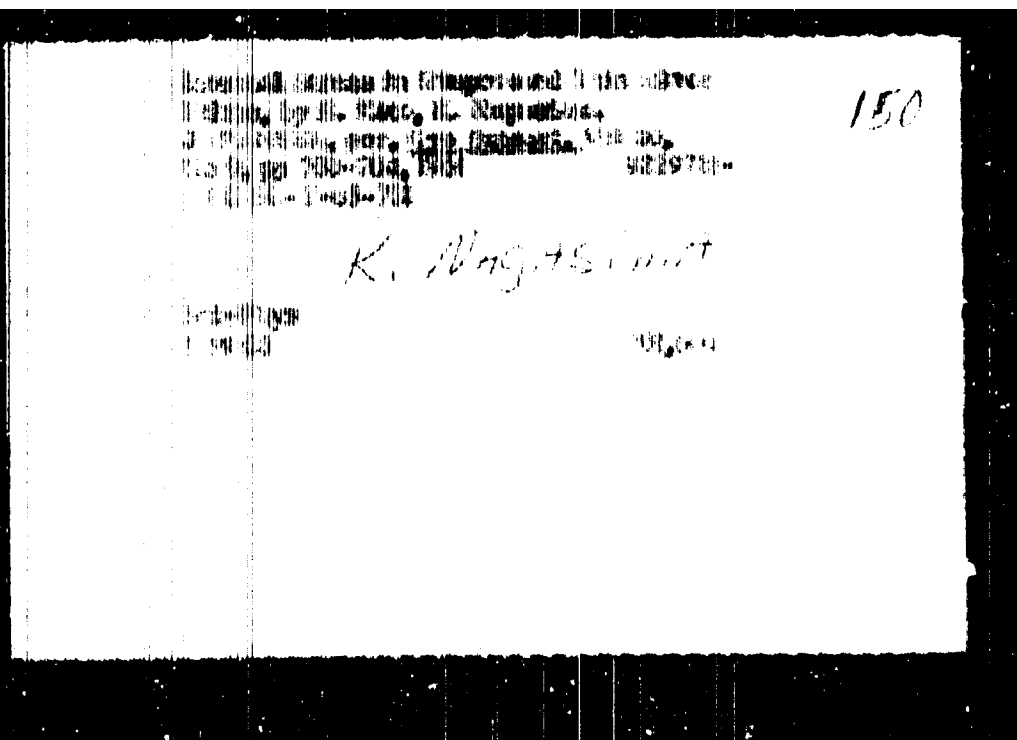
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